

## Structure of the Polyether Antibiotic K-41 by Mass and Nuclear Magnetic Resonance Spectrometry

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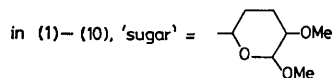
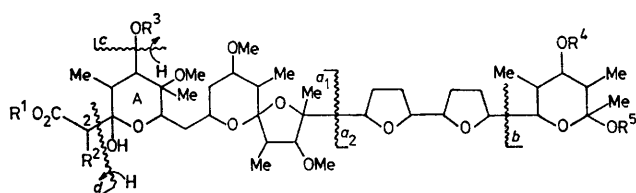
**Summary** Using electron impact and field desorption mass spectrometry and n.m.r. spectrometry, the polyether antibiotic K-41 has been assigned structure (1).

THE accompanying communication<sup>1</sup> describes determination of the structure of the antibiotic K-41<sup>2</sup> using X-ray diffraction. By personal communication, we found that A32887, an antibiotic produced at these laboratories by *Streptomyces albus* NRRL 11109, was identical to K-41. We outline here independent and concurrent work leading to our proposal of the structure (1) for K-41 (A32887).

The molecular formula  $C_{48}H_{82}O_{18}$  for (1) was established by accurate mass measurement of the molecular ion ( $M^+$ )<sup>†</sup> in its field desorption (F.D.) spectrum and by microanalysis. A similar formula,  $C_{49}H_{84}O_{17}$ , for A204A (2), a polyether antibiotic of known structure,<sup>3</sup> prompted us to compare the n.m.r. and mass spectra of these two compounds in detail.

The <sup>13</sup>C- and <sup>1</sup>H-n.m.r. spectra of the sodium salt of (1) are similar to the spectra of the sodium salt of (2). In particular, the n.m.r. spectra indicate the presence of the sugar unit and the spiroacetal system in (1). However, the quartet in the <sup>1</sup>H spectrum<sup>4</sup> of the sodium salt of (2) due to the hydrogen

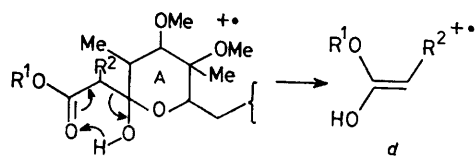
<sup>†</sup> Two formulae,  $C_{48}H_{82}O_{18}Na$  and  $C_{55}H_{78}O_{15}Na$ , give peaks within the range  $969.5373 \pm 0.010$  found by peak matching with a field ionized reference standard of hexakis-(2,2,3,3-tetrafluoro-propoxy)cyclotriphosphazene at resolution 8000. The latter formula was excluded by microanalysis.



	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	R <sup>4</sup>	R <sup>5</sup>	
(1)	H	OH	Me	Sugar	H	(K-41)
(2)	H	Me	Sugar	Me	H	(A204A)
(3)	Me	Me	Sugar	Me	H	
(4)	Me	OH	Me	Sugar	H	
(5)	H	OAc	Me	Sugar	H	
(6)	Ph <sub>2</sub> CH	OH	Me	Sugar	Me	
(7)	Me	OCOPh	Me	Sugar	Me	
(8)	Me	OH	Me	Sugar	Me	
(9)	Me	OH	Me	Sugar	CD <sub>3</sub>	
(10)	Me	Me	Sugar	Me	Me	

at C-2 is missing in the <sup>1</sup>H spectrum of the sodium salt of (1). A prominent peak at *m/e* 129 (C<sub>7</sub>H<sub>13</sub>O<sub>2</sub>) in the electron impact (E.I.) and F.D. spectra of (1) and (2) supports the presence of the sugar unit in (1).

The E.I. and F.D. spectra of the methyl ester (3) of A204A(2) show a prominent peak *a*<sub>1</sub> at *m/e* 645. In the E.I. spectrum, *a*<sub>1</sub> fragments by elimination of the molecules *c* and *d* to give the ion C<sub>22</sub>H<sub>35</sub>O<sub>7</sub> which further fragments by the successive elimination of three molecules of methanol. The E.I. and F.D. spectra of the methyl ester (4) of K-41 (1) show no *m/e* 645 peak but instead a peak at *m/e* 533. In the E.I. spectrum, an ion C<sub>23</sub>H<sub>39</sub>O<sub>8</sub> fragments by the successive elimination of 4 molecules of methanol. All the foregoing data can be accommodated by having R<sup>2</sup> = OH and R<sup>3</sup> = Me in the structure of (1); the elemental composition of *m/e* 533 (*a*<sub>1</sub>) supports this conclusion. Formation of an acetate (5) by (1) but not by (2) indicates the presence of an additional hydroxy in (1).

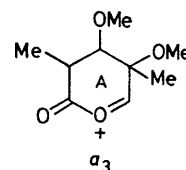


SCHEME

(6) *m/e* 242, C<sub>16</sub>H<sub>14</sub>O<sub>3</sub>  
 (7) *m/e* 194, C<sub>10</sub>H<sub>10</sub>O<sub>4</sub>

Definitive evidence for the presence of the unit CO<sub>2</sub>H–CH(OH)– in (1) comes from the E.I. spectra of its diphenylmethyl ester methyl ether (6) and methyl ester methyl ether benzoate (7). The E.I. spectra of both (6) and (7) show the fragmentation in the Scheme. That the itinerant hydrogen involved in the McLafferty rearrangement is an exchangeable one is shown by a 1 a.m.u. shift to higher mass of the ions *d* when the E.I. spectra of (6) and (7) exchanged with CH<sub>3</sub>OD are determined. The presence of the hydroxy-group  $\alpha$  to the carboxylic acid group of (1) results in its *pK<sub>a</sub>* being two units lower than the *pK<sub>a</sub>* of (2).

The E.I. spectra of (1) or its derivatives yield no discernible evidence regarding the units *a*<sub>2</sub> or *b*. However, the F.D. spectrum of the methyl ester methyl ether (8) of (1), taken somewhat above the best anode temperature, yields the peaks *a*<sub>2</sub> and *b* having the correct elemental composition. In the F.D. spectrum of the methyl ester deuteriomethyl ether (9) of (1) the peaks *a*<sub>2</sub> and *b* are shifted 3 a.m.u. to higher mass. Analogous fragments were observed in the F.D. spectrum of the methyl ester methyl ether (10) of (2).



A fragment C<sub>9</sub>H<sub>15</sub>O<sub>4</sub><sup>+</sup> to which we assign the structure *a*<sub>3</sub> occurs in the E.I. spectra of compounds (7) and (9). It arises from fragment *a*<sub>1</sub>, without molecule *d* and two molecules of methanol, and most likely represents ring (A) rather than another portion of (1).

Although the foregoing data do not result in a rigorous structure determination of such a relatively complex molecule as (1), they were sufficient, taken in conjunction with the knowledge of the structures of similar compounds, to lead to a correct structure proposal. The application of these methods to the structure determination of metabolites of polyethers of this type is obvious.

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<sup>1</sup> M. Shiro, H. Nakai, K. Nagashima, and N. Tsuji, *J.C.S. Chem. Comm.*, preceding communication.

<sup>2</sup> N. Tsuji, K. Nagashima, M. Kobayashi, Y. Wakisaka, Y. Kawamura, S. Kōzuki, and M. Mayama, *J. Antibiotics*, 1976, **29**, 10.

<sup>3</sup> N. D. Jones, M. O. Chaney, J. W. Chamberlin, R. L. Hamill, and S. Chen, *J. Amer. Chem. Soc.*, 1973, **95**, 3399.

<sup>4</sup> M. J. O. Anteunis and G. Verhege, *Bull. Soc. chim. Belges*, 1977, **86**, 355.